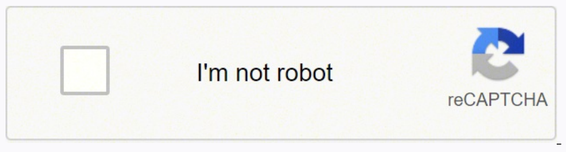

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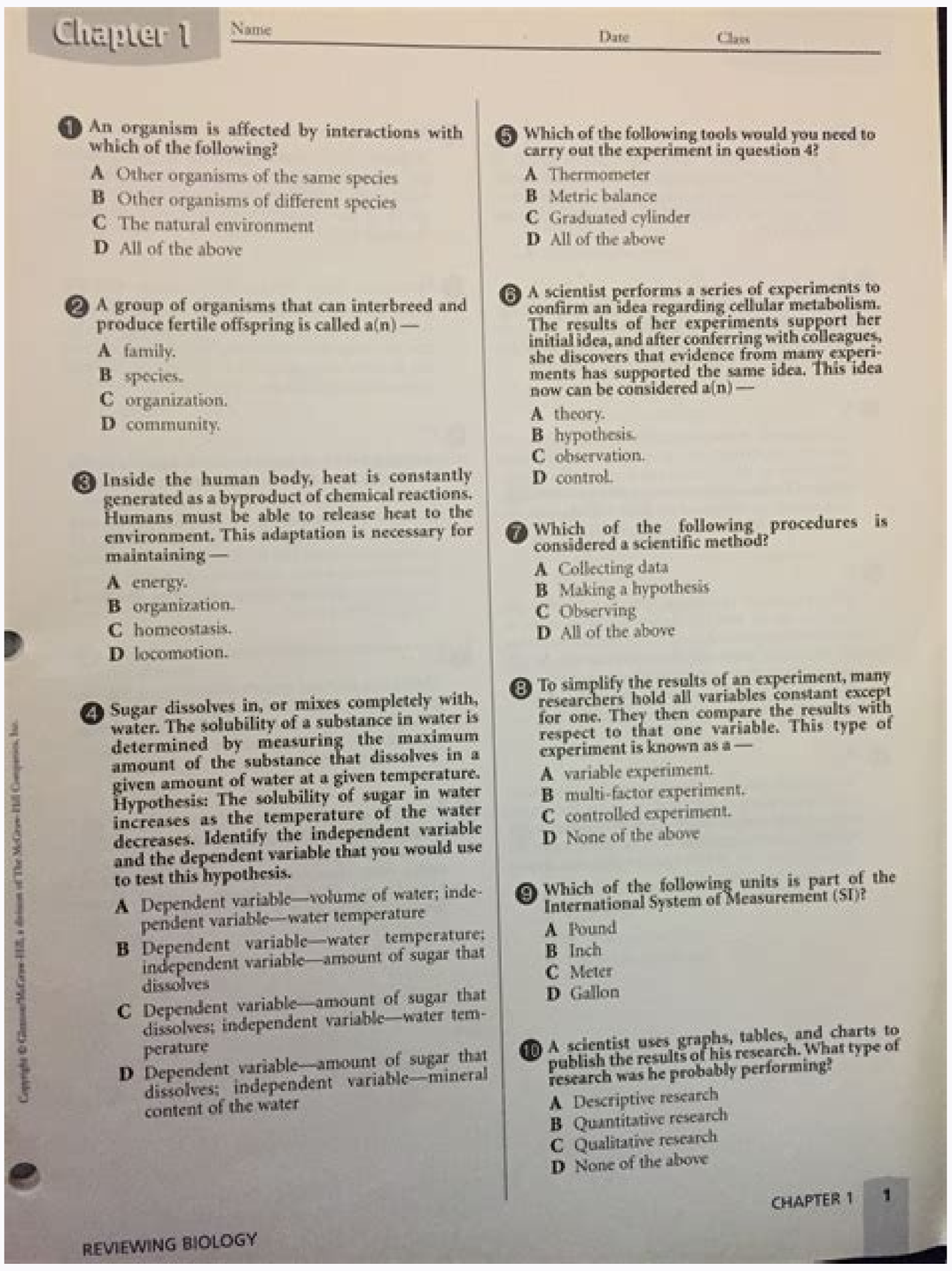

Chapter 1 Name Date Class

1 An organism is affected by interactions with which of the following?
A Other organisms of the same species
B Other organisms of different species
C The natural environment
D All of the above

2 A group of organisms that can interbreed and produce fertile offspring is called a(n) —
A family.
B species.
C organization.
D community.

3 Inside the human body, heat is constantly generated as a byproduct of chemical reactions. Humans must be able to release heat to the environment. This adaptation is necessary for maintaining —
A energy.
B organization.
C homeostasis.
D locomotion.

4 Sugar dissolves in, or mixes completely with, water. The solubility of a substance in water is determined by measuring the maximum amount of the substance that dissolves in a given amount of water at a given temperature. Hypothesis: The solubility of sugar in water increases as the temperature of the water decreases. Identify the independent variable and the dependent variable that you would use to test this hypothesis.
A Dependent variable—volume of water; independent variable—water temperature
B Dependent variable—water temperature; independent variable—amount of sugar that dissolves
C Dependent variable—amount of sugar that dissolves; independent variable—water temperature
D Dependent variable—amount of sugar that dissolves; independent variable—mineral content of the water

5 Which of the following tools would you need to carry out the experiment in question 4?
A Thermometer
B Metric balance
C Graduated cylinder
D All of the above

6 A scientist performs a series of experiments to confirm an idea regarding cellular metabolism. The results of her experiments support her initial idea, and after conferring with colleagues, she discovers that evidence from many experiments has supported the same idea. This idea now can be considered a(n) —
A theory.
B hypothesis.
C observation.
D control.

7 Which of the following procedures is considered a scientific method?
A Collecting data
B Making a hypothesis
C Observing
D All of the above

8 To simplify the results of an experiment, many researchers hold all variables constant except for one. They then compare the results with respect to that one variable. This type of experiment is known as a —
A variable experiment.
B multi-factor experiment.
C controlled experiment.
D None of the above

9 Which of the following units is part of the International System of Measurement (SI)?
A Pound
B Inch
C Meter
D Gallon

10 A scientist uses graphs, tables, and charts to publish the results of his research. What type of research was he probably performing?
A Descriptive research
B Quantitative research
C Qualitative research
D None of the above

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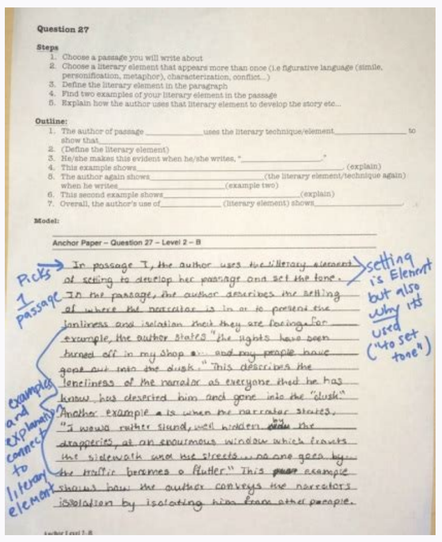


Question 27

Steps
1. Choose a passage you will write about
2. Choose a literary element that appears more than once (i.e figurative language (simile, personification, metaphor), characterization, conflict...)
3. Define the literary element in the paragraph
4. Find two examples of your literary element in the passage
5. Explain how the author uses that literary element to develop the story etc...

Outline:
1. The author of passage _____ uses the literary technique/element_____ to show that_____
2. (Define the literary element)
3. He/she makes this evident when he/she writes, "_____"
4. This example shows_____ (explain)
5. The author again shows_____ (the literary element/technique again) when he writes_____ (example two)
6. This second example shows_____ (explain)
7. Overall, the author's use of_____ (literary element) shows_____

Model:

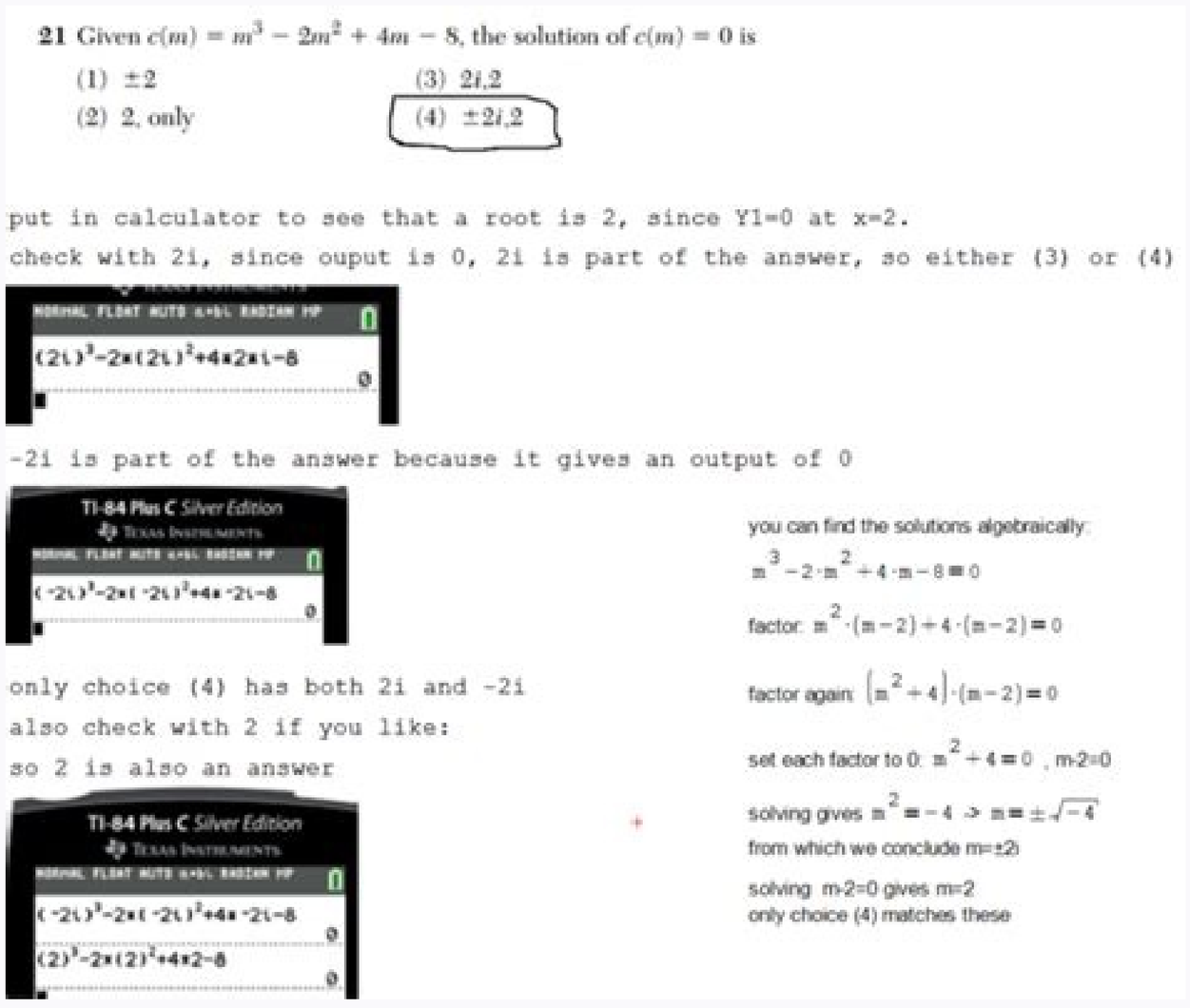
Anchor Paper – Question 27 – Level 2 – B

Picks 1 passage

In passage 1, the author uses the literary element of setting to develop her passage and set the tone. In her passage, the author describes the setting of where the narrator is in order to present the lonliness and isolation that they are feeling. For example, the author states "the lights have been turned off in my shop ... and my people have gone out into the dusk." This describes the loneliness of the narrator as everyone that he has known has deserted him and gone into the "dusk". Another example is when the narrator states, "I would rather stand, with wisdom at the shuttered, at an enourmous window which frames the sidewalk and the streets ... no one goes by. the traffic becomes a flutter." This ~~poor~~ example shows how the author conveys the narrators isolation by isolating him from other people.

Setting is Element but also why it is used ("to set tone")

Example and explanation connect to literary element



21 Given $c(m) = m^3 - 2m^2 + 4m - 8$, the solution of $c(m) = 0$ is

(1) ± 2
(2) 2, only
(3) $2i, 2$
(4) $\pm 2i, 2$

put in calculator to see that a root is 2, since Y1=0 at x=2.
check with 2i, since ouput is 0, 2i is part of the answer, so either (3) or (4)

-2i is part of the answer because it gives an output of 0

only choice (4) has both 2i and -2i
also check with 2 if you like:
so 2 is also an answer

you can find the solutions algebraically:

$m^3 - 2 \cdot m^2 + 4 \cdot m - 8 = 0$

factor: $m^2 \cdot (m-2) + 4 \cdot (m-2) = 0$

factor again: $\left(m^2 + 4\right) \cdot (m-2) = 0$

set each factor to 0: $m^2 + 4 = 0$, m-2=0

solving gives $m^2 = -4 \rightarrow m = \pm\sqrt{-4}$

from which we conclude m=±2)

solving m-2=0 gives m=2
only choice (4) matches these

Part A

Answer all questions in this part.

Directions (1–30): For *each* statement or question, record on your separate answer sheet the *number* of the word or expression that, of those given, best completes the statement or answers the question. Some questions may require the use of the *2011 Edition Reference Tables for Physical Setting/Chemistry*.

1 Compared to the charge of a proton, the charge of an electron has
(1) a greater magnitude and the same sign
(2) a greater magnitude and the opposite sign
(3) the same magnitude and the same sign
(4) the same magnitude and the opposite sign

2 Which atom has the largest atomic radius?
(1) potassium
(2) rubidium
(3) francium
(4) cesium

3 In the wave-mechanical model of the atom, an orbital is defined as
(1) a region of the most probable proton location
(2) a region of the most probable electron location
(3) a circular path traveled by a proton around the nucleus
(4) a circular path traveled by an electron around the nucleus

4 When an excited electron in an atom moves to the ground state, the electron
(1) absorbs energy as it moves to a higher energy state
(2) absorbs energy as it moves to a lower energy state
(3) emits energy as it moves to a higher energy state
(4) emits energy as it moves to a lower energy state

5 Which polyatomic ion is found in the compound represented by the formula $NaHCO_3$?
(1) acetate
(2) hydrogen carbonate
(3) hydrogen sulfate
(4) oxalate

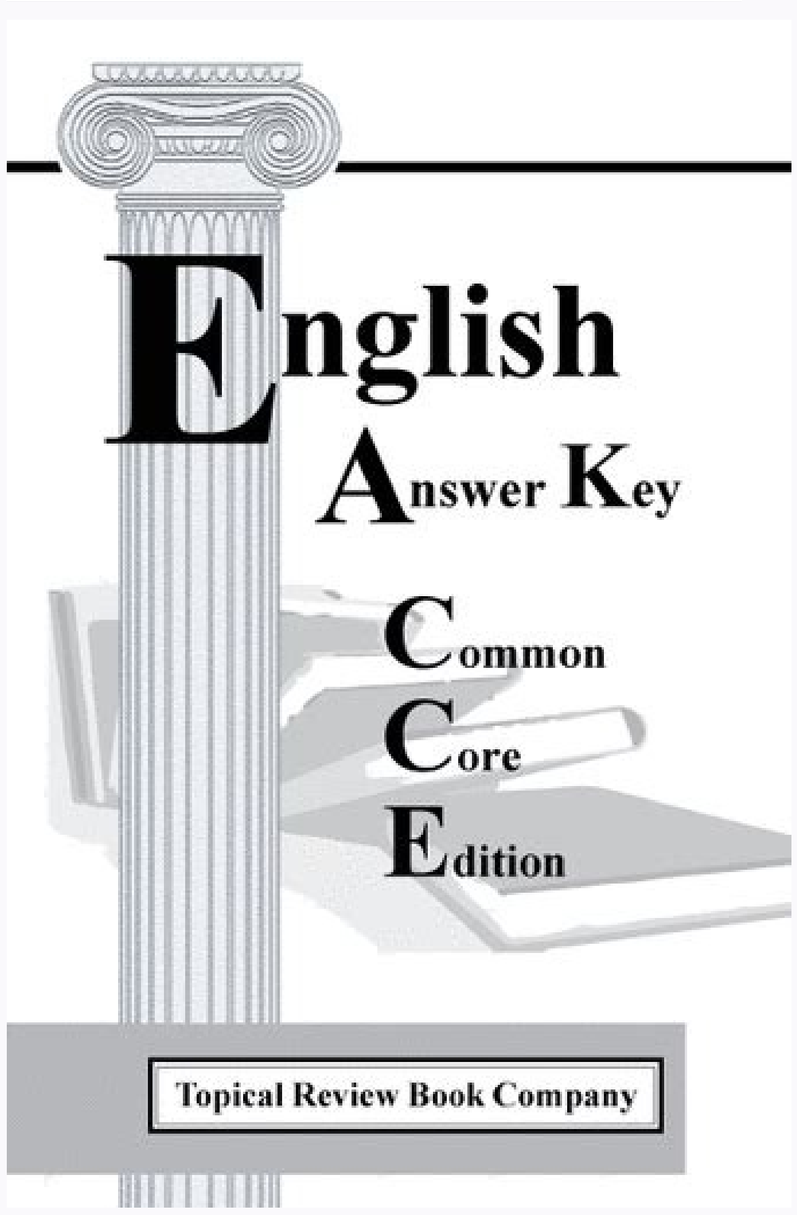
6 The atomic mass of magnesium is the weighted average of the atomic masses of
(1) all of the artificially produced isotopes of Mg
(2) all of the naturally occurring isotopes of Mg
(3) the two most abundant artificially produced isotopes of Mg
(4) the two most abundant naturally occurring isotopes of Mg

7 Which element has atoms that can form halide ions?
(1) iodine
(2) silver
(3) strontium
(4) xenon

8 Two forms of solid carbon, diamond and graphite, differ in their physical properties due to the differences in their
(1) atomic numbers
(2) crystal structures
(3) isotopic abundances
(4) percent compositions

9 Which quantity can be calculated for a solid compound, given only the formula of the compound and the Periodic Table of the Elements?
(1) the density of the compound
(2) the heat of fusion of the compound
(3) the melting point of each element in the compound
(4) the percent composition by mass of each element in the compound

10 Which terms identify types of chemical reactions?
(1) decomposition and sublimation
(2) decomposition and synthesis
(3) deposition and sublimation
(4) deposition and synthesis



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